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IN THE UNITED STATES PATENT AND TRADEMARK OFFICE  
UTILITY PATENT APPLICATION TRANSMITTAL LETTER

Attorney Docket No.: CT00-013

Mailing Date: August 28, 2000

Express Mail #EL568745916US

To: Assistant Commissioner for Patents  
Box Patent Application  
Washington D.C., 20231

Dear Sir:

Transmitted herewith for filing under 37 C.F.R. 1.53(b) is a:

- ☒ New Nonprovisional Utility Patent Application; or
- ☐ Continuation; or ☐ Divisional; or ☐ Continuation-In-Part (CIP);  
of prior US Application No. \_\_\_\_\_, filed on \_\_\_\_\_, having U.S.  
Examiner \_\_\_\_\_, in Group Art Unit \_\_\_\_\_

Of: Chowdary Ramesh Koripella, Christopher K. Dyer, Stephen P. Rogers, and Dominic Francis Gervasio

For: **HYDROGEN GENERATOR UTILIZING CERAMIC TECHNOLOGY**

- ☒ 6 sheets of informal drawings and 28 pages of specification and claims.
- ☒ Newly executed oath or declaration combined with Power of Attorney on 3 pages.
- ☐ Copy of oath or declaration from prior U.S. application serial no. \_\_\_\_\_  
☐ The following named inventor(s) from the prior application are hereby deleted from this application in accordance with 37 C.F.R. 1.63(d)(2) and 1.33(b):  
\_\_\_\_\_
- ☐ Foreign priority to EPO patent application having serial number \_\_\_\_\_ and a filing date of \_\_\_\_\_, is hereby claimed under 35 USC 119.
- ☒ An Assignment Transmittal Letter and Assignment of the invention to Motorola, Inc.
- ☐ An Information Disclosure Statement (IDS), with PTO-1449, and \_\_\_\_\_ citation copies.
- ☒ Return Receipt Postcard.
- ☐ Preliminary Amendment.
- ☐ Please cancel pending claims \_\_\_\_\_.
- ☐ Incorporation by Reference (for Continuation/Division/CIP application). The entire disclosure of the prior application, from which a copy of the oath or declaration is supplied, is considered as being part of the disclosure of the accompanying application and is hereby incorporated by reference therein. Since the present application is based on a prior US application, please amend the specification by adding the following sentence before the first sentence of the specification:
- "The present application is based on prior US application No. \_\_\_\_\_, filed on \_\_\_\_\_, which is hereby incorporated by reference, and priority thereto for common subject matter is hereby claimed."
- ☐ Applicant hereby petitions pursuant to 37 C.F.R. § 1.136(a) for a \_\_\_\_\_ month extension of time for response to the outstanding Official Action mailed \_\_\_\_\_. The period for response was previously set to elapse \_\_\_\_\_, and is accordingly hereby extended to \_\_\_\_\_, which is still within the six-month statutory period for response (35 U.S.C. § 133) which elapses \_\_\_\_\_. The reason for this petition is that a Division, Continuation, or CIP is being filed, and it is desired to maintain the present application in pending condition pursuant to 35 USC § 120 through at least the filing of the Division, Continuation, or CIP application. The required Extension Fee established by 37 C.F.R. § 1.17(a) pursuant to 35 U.S.C. § 41(a) (8) is:

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09/04/00  
09/04/00

EXTENSION	FEE
<input type="checkbox"/> First Month	\$110.00
<input type="checkbox"/> Second Month	\$380.00
<input type="checkbox"/> Third Month	\$870.00
<input type="checkbox"/> Fourth Month	\$1,360.00
<input type="checkbox"/> Fifth Month	\$1,850.00

- ☒ The filing fee is calculated as follows:

CLAIMS AS FILED, LESS ANY CANCELED BY AMENDMENT

FOR	NUMBER OF CLAIMS	NUMBER EXTRA	RATE	FEE
TOTAL CLAIMS	21 - 20 =	1	x \$18	= \$ 18.00
INDEPENDENT CLAIMS	3 - 3 =	0	x \$78	= \$ 0.00
MULTIPLE DEPENDENT CLAIMS			\$260	= \$ 0.00
BASIC FEE				= \$ 690.00
TOTAL FILING FEE				= \$ 708.00

- ☒ Please charge Deposit Account No. 13-4771 in the amount of \$ 708.00 for the Total Filing Fee, and the Extension Fee under 37 C.F.R. §1.136(a), if applicable.
- ☒ The Commissioner is hereby authorized to charge any additional fees which may be required now or in the future during the entire pendency of this application under 37 C.F.R. 1.16 or 37 C.F.R. 1.17, including any present or future time extension fees which may be required, or credit any overpayment to Deposit Account No. 13-4771.
- ☒ This sheet is submitted in duplicate.

This transmittal letter has 2 total pages.

28 Aug 2000  
DATE

William E. Koch 29,659  
William E. Koch REG. NO.

Motorola, Inc.  
Customer Number: 29330

Attorney of Record  
Telephone No.: (480) 441-4281  
Facsimile No.: (480) 441-5220

HYDROGEN GENERATOR UTILIZING CERAMIC TECHNOLOGY

Field of Invention

The present invention pertains to hydrogen generators, and more particularly to hydrogen generators utilizing multi-layer ceramic technology for improved size and performance benefits.

Background of the Invention

Hydrogen is attractive and has been used as the fuel in many types of devices, such as: in fuel cells for producing electricity, in combustion engines, including rockets or internal combustion engines, and in chemical processing devices, including those used in the refining of metallic components. Typically hydrogen, or more particularly, hydrogen enriched gas, is processed from methanol, natural gas, petroleum, or ammonia using a chemical reactor, typically called a hydrogen generator or reformer. A reformed hydrogen fuel cell (RHFC), utilizes

hydrogen enriched gas processed from liquid or gaseous hydrocarbon fuels, such as methanol, using a fuel reformer.

Methanol is the preferred fuel for use in fuel reformers, or hydrogen generators, because methanol is easier to reform into hydrogen enriched gas at a relatively low temperature and yields a minimum number of gaseous by-products (carbon dioxide and carbon monoxide) compared to other hydrocarbon fuels such as natural gas, ethanol, petroleum, gasoline, or butane. This is especially important for small portable units, where the temperature of the reformer unit would be a concern. The reforming or converting of methanol into hydrogen enriched gas usually is accomplished using one of three different types of reforming processes. These three types are steam reforming, partial oxidation reforming, and autothermal reforming. Of these types, steam reforming is the preferred process for methanol reforming because it is the easiest to control, yields the minimum number of gaseous by-products (carbon dioxide and carbon monoxide), and produces a higher hydrogen output, at a lower temperature, thus lending itself to favored use. During steam reforming, raw methanol is catalytically converted, in the

presence of water and with the application of heat, to a hydrogen enriched gas. Although, steam reforming is the preferred process, partial oxidation reforming and autothermal reforming are utilized in many instances. During partial oxidation reforming, raw methanol is converted to a hydrogen enriched gas through partial oxidation of methanol over a catalyst in a limited supply of oxygen or air to prevent complete oxidation. Since this is an exothermic reaction, it does not require heat input to proceed. The reaction will proceed without any additional heat input provided the methanol and oxygen are in contact with the proper catalyst. Control of oxygen partial pressure and temperature is very critical, and for portable methanol reforming, the higher operating temperature of the reformer is a concern.

Autothermal reforming is a combination of the catalytic partial oxidation and steam reforming process. During the autothermal methanol reforming process, the partial oxidation reaction which produces heat is carefully managed to provide sufficient heat for the steam reforming reaction. In an autothermal reformer, the reactions between the input reactants, namely the methanol, water and air(or

$O_2$ ), are carefully balanced over the catalyst, to produce  $CO_2$  and  $H_2$  gases with minimum amount of  $CO$ . The partial oxidation step and steam reforming step may be done in the same or separate compartments during autothermal reforming.

Fuel reformers have been developed for use in conjunction with many new devices, including fuel cell devices. Many of these fuel cell devices include reformers which are typically cumbersome and complex devices consisting of several discrete sections connected together with gas plumbing and hardware to produce hydrogen gas. Accordingly, reformers have not been found suitable for portable power source applications, or in other applications requiring minimal size and weight. To date, no fuel reformer has been developed utilizing ceramic monolithic structures in which the miniaturization of the reformer has been achieved. Laminated ceramic components containing miniature channels and other features which utilize low pressure lamination ceramic technology, are now commonly being developed for use in microfluidic management systems. Monolithic structures formed of these laminated ceramic components provide for three-dimensional structures that are inert and stable to chemical reactions and capable

of tolerating high temperatures as well as providing for miniaturized structures, with a high degree of electronic circuitry or components embedded or integrated into such a ceramic structure for system control and functionality. Additionally, the ceramic materials used to form ceramic devices, which have microchannels formed within the structure, are considered to be excellent candidates for catalyst supports in microreactor devices for generating hydrogen for supplying miniaturized fuel cells.

Accordingly, it is an object of the present invention to provide for a miniaturized hydrogen generator, or fuel processor, that provides for the reforming of a fuel to a hydrogen enriched gas.

It is yet another object of the present invention to provide for a monolithic structure for the reforming of a fuel to a hydrogen enriched gas.

It is still another object of the present invention to provide for a monolithic structure that is formed by utilizing ceramic technology, thereby providing for the integration of a plurality of internal plumbing interconnections and electrical circuitry and connections.

It is another object of the present invention to provide for a hydrogen generator, or fuel processor, that is miniaturized for use in conjunction with: (i) fuel cells for portable device applications; (ii) combustion devices; (iii) chemical processing devices; and (iv) other devices in which hydrogen enriched gas is consumed as fuel.

#### Summary of the Invention

The above problems and others are at least partially solved and the above purposes and others are realized in a hydrogen generator including a three-dimensional multilayer ceramic carrier structure defining a fuel reformer, or fuel processor. The fuel reformer includes a vaporization zone and a reaction zone including a reforming catalyst. The ceramic carrier further includes an inlet channel for liquid fuel and an outlet channel for the hydrogen enriched gas. The fuel reformer is formed utilizing ceramic technology in which thin ceramic layers are assembled then sintered to provide miniature dimensions in which the



encapsulated catalyst converts or reforms inlet fuel into a hydrogen enriched gas.

### Brief Description of the Drawings

The novel features believed characteristic of the invention are set forth in the claims. The invention itself, however, as well as other features and advantages thereof will be best understood by reference to detailed descriptions which follow, when read in conjunction with the accompanying drawings, wherein:

FIG. 1 is a simplified orthogonal view of a hydrogen generator including a plurality of channels, according to the present invention;

FIG. 2 is a simplified orthogonal view of an alternative embodiment of the hydrogen generator according to the present invention;

FIG. 3 is a simplified orthogonal view of an alternative embodiment of the hydrogen generator according to the present invention;

FIG. 4 is a simplified sectional views of a thermally conductive channels or vias according to the present invention;

FIGs. 5 and 6 are simplified sectional views of alternative embodiments of thermally controlled vaporizer/reaction zone channels according to the present invention; and

FIG. 7 is a simplified flow diagram of a method of reforming fuel according to the present invention.

#### Description of the Preferred Embodiments

Turning now to the drawings, and in particular FIG. 1, illustrated in simplified orthogonal view is a hydrogen generator 10 including a plurality of channels, according to the present invention. Hydrogen generator 10 is comprised of a three-dimensional multi-layer ceramic structure 12. Ceramic structure 12 is formed utilizing multi-layer laminate ceramic technology. Structure 12 is typically formed in component parts which are then sintered in such a way as to provide for a monolithic structure.

Ceramic structure 12 has defined therein a fuel reformer, or fuel processor, generally referenced 14. Fuel reformer 14 includes a vaporization chamber, or vaporization zone, 16, a reaction chamber, or reaction zone, 18, and an integrated heat source 28. Ceramic structure 12 further includes at least one fuel inlet channel 20 in communication with a liquid fuel source of a combination solution methanol/water source 24 and a pure methanol source 26 and a hydrogen enriched fuel outlet channel 22. It should be understood that in a different embodiment, separate methanol and pure water fuel sources can be used. In this instance the methanol would need to be in direct communication with the heater and a mixer would need to be utilized for mixing of the methanol and water fuel source.

During operation, heat is efficiently transferred from the central aspect of the device, more particularly an integrated heater 28, to the reaction zone 18 and fuel vaporizer, or vaporization zone 16 using thermal conductive channels, or vias, (discussed presently). In this particular embodiment, integrated heater 28 is described as a chemical heater, including a catalyst and arranged so as to oxidize fuel to produce heat, but it should be

understood that the integration of an electrical heater is anticipated by this disclosure. Chemical heater 28 includes an air port 40 for providing oxygen for oxidation of fuel 24 and/or 26 and an inlet channel 20, for providing fuel 24 and/or 26 to heater 28.

Output from fuel vaporizer zone 16 travels via structure 30 to reaction zone 18 and then through hydrogen enriched gas outlet channel 22. Optionally included is a preferential oxidation reactor 21. During operation, output from reaction zone 18 travels to preferential oxidation reactor 21 where CO is preferentially cleaned up from the output gasses to below 30 ppm for fuel cell use. Alternatively, when a high temperature fuel cell is utilized, design 10 would not include preferential oxidation reactor 21 due to the ability for high temperature fuel cells to tolerate up to a combined total of 5% (mol) CO and methanol. Additionally, included in design 10 is a waste heat recovery zone 34 for the capture of heat from spent gases from an external device, such as a fuel cell stack, (not shown) through an inlet channel (not shown) and from chemical heater 28 through channel 27.

An efficient thermal insulator 36 is positioned about fuel reformer 14 to keep outer temperatures low for packaging and also to keep heat localized to the fuel reformer. In this particular embodiment, the fuel processor operates at a temperature ranging from ambient to 300°C unless it is integrated with a high temperature fuel cell where the fuel processor operates in a range of 140-300°C. Fuel vaporizer zone 16 operates at a temperature ranging from 120-160°C and reaction zone 18 operates at a temperature ranging from 200-250°C. Additionally, in this particular embodiment of fuel processor 10, included is an exhaust gas vent 38, for the venting of exhaust gases generated by device 10.

It should be understood that alternative embodiments encompassing alternative fuel delivery means and fuel vaporizer, reaction zone, and chemical heater positions are anticipated by this disclosure. In particular, anticipated is an embodiment in which only a single fuel supply, namely methanol and water, is anticipated. This use of a single methanol and water solution would enable the fabrication of a simpler design, without any need for the device to incorporate two fuel tanks. Although it is understood that

pure methanol is more efficient and preferred with respect to the chemical heater, a 1 mole water and 1 mole methanol solution will also work, but is not deemed literally as operationally efficient. Further, a heater using the water and methanol solution is suitable for practical applications, and would permit a simple common fuel reservoir for feeding the heater and reformer chambers. In this instance, the fuel delivery would be split into two chambers, the chemical heater 28 and fuel vaporizer 16.

Next, anticipated are variations on the actual design of device 10 and more particularly to the actual location of the fuel vaporizer zone 16, reaction zone 18 and chemical heater 28. In one particular alternative embodiment, it is anticipated that reaction zone 18 surrounds chemical heater 28 on both sides (top and bottom). In yet another alternative embodiment, it is anticipated that reaction zone 18 can be positioned below heater 28 and the fuel vaporizer zone 16 on top of the chemical heater 28. In addition, and as previously stated, it is anticipated that chemical heater 28 can alternatively be an electrical heater (not shown).

Finally, it is anticipated by this disclosure that although illustrated in FIG. 1 is a design in which a fuel cell is not integrated with reformer 14, it is additionally anticipated, that a fuel cell (not shown) can be monolithically integrated with hydrogen generator 10. Further information on a reformed hydrogen fuel system device including an integrated fuel cell stack can be found in U.S. Patent application, bearing attorney docket number CT 00-014, simultaneously filed herewith, entitled "FUEL PROCESSOR WITH INTEGRATED FUEL CELL UTILIZING CERAMIC TECHNOLOGY", assigned to the same assignee and incorporated herein by this reference.

Referring now to FIG. 2, illustrated is a partial oxidation hydrogen generator according to the present invention, generally referenced 50. Partial oxidation reformer 50 provides for part of the fuel to be oxidized to provide energy for the reforming reaction within the fuel reformer. Direct heat transfer, makes this reformer small, compact, light weight and dynamically responsive. Partial oxidation systems rely on the reaction of the feedstock in a limited supply of oxygen or air to prevent complete oxidation. Partial oxidation is an exothermic reaction and

in this process, the temperature of the reformer can exceed 400°C. Control of the oxygen partial pressure is very critical. Partial oxidation is not typically used for methanol reformers, and is more commonly utilized in association with gasoline reformers for automotive applications where the temperatures can exceed 650°C.

As illustrated, hydrogen generator 50 includes a fuel 52, input through inlet 54, to a partial oxidation reactor 56. During operation, fuel 52, typically methanol, is first oxidized by a chemical heater or electrical igniter, 58 over a catalyst. Once the partial oxidation reaction starts, the heat generated will sustain the reaction. There is no need to provide additional heat as in the steam reformer described with reference to FIG. 1. During operation, the reaction needs to be controlled so that it does not proceed to complete oxidation. This is achieved by monitoring the oxygen partial pressure and temperature and by controlling the air intake 60 and fuel feed 54.

Fuel 52 travels via inlet 54 to fuel reformer 68, more particularly to first partial oxidation reactor 56 and then via channel 62, to a second partial oxidation reactor 64. An enriched hydrogen fuel outlet 66 serves to direct the



reformed hydrogen fuel toward an externally connected fuel cell (not shown).

An efficient thermal insulator 70 is positioned about partial oxidation reactors 56 and 64 to keep outer temperatures low for packaging and also to keep heat localized to the fuel reformer. It is anticipated by this disclosure that although illustrated in FIG. 2 is a design in which a fuel cell is not integrated with reformer 68, it is additionally anticipated, that a fuel cell (not shown) can be monolithically integrated with hydrogen generator 50. In this particular design, hydrogen generator 50 includes a temperature sensor 72 and an oxygen partial pressure,  $PO_2$ , sensor 74.

As illustrated in FIG. 2, partial oxidation hydrogen generator 50 converts oxygen and methanol mainly into hydrogen and carbon dioxide with a small amount of carbon monoxide. Since the partial oxidation reaction is an exothermic reaction, this reforming reaction does not require the addition of heat. The reforming is performed over a temperature range of  $200^{\circ}\text{C}$  -  $400^{\circ}\text{C}$ . CO cleanup is achieved in much the same way as was done with the steam reformer illustrated in FIG. 1.

Referring now to FIG. 3, illustrated is an autothermal hydrogen generator according to the present invention, generally referenced 80. Autothermal reforming is a combination of both steam reforming, an endothermic reaction which requires heat input described with reference to FIG. 1, and partial oxidation reforming, an exothermic reaction which produces heat, described with reference to FIG. 2. In autothermal reforming, a balance of both of these reforming processes is achieved. More particularly, autothermal reforming converts fuel 82, more particularly water, oxygen and methanol, to hydrogen and carbon dioxide without the requirement of the addition of heat.

Fuel 82 travels via inlet 81 to a fuel reformer 83, more particularly to the first stage of the autothermal reactor 87 involving partial oxidation of some of the methanol. The fuel stream is interconnected via channel 88, to a second stage of the autothermal reactor 90 involving steam reforming of the methanol. These two reforming stages may occur either in the same compartment or in two sequential separate compartments. Either way, an enriched hydrogen fuel outlet 92 serves to direct the

reformed hydrogen fuel toward an externally connected fuel cell (not shown).

Temperature sensor 84 and an oxygen partial pressure sensor 85 are utilized to monitor the reaction process. In particular, sensors 84 and 85 control the fuel feed (methanol and water), 82, and oxygen supply 86. If there is too much oxygen present then the reaction can proceed towards complete oxidation, consuming excess fuel, and producing unwanted excessive heat. In contrast, if sufficient heat is not produced due to partial oxidation reaction, then coking can result, hindering the steam reforming process. Autothermal methanol reforming proceeds in the temperature range of 200-300°C with the use of proper catalyst(s). CO cleanup is achieved in much the same way as was discussed with the steam reformer illustrated in FIG. 1.

Referring now to FIG. 4, illustrated in simplified cross-sectional view is a thermal transfer structure, referenced 30. Structure 30 is comprised of channels 102 or vias connected with a metal film 104 in intimate contact with the various zones 18 and 28, as illustrated. It should be understood that structure 30 is generally

utilized for transfer of heat efficiently between chemical heater 28 and fuel vaporizer zone 16 and reaction zone 18, as well as between waste heat recovery 34 and reaction zone 18, or any other zone where thermal transfer needs to occur. As illustrated, thermal conductive structure 30 includes multi-layer ceramic layers 100 used in the fabrication of the monolithic structure, referenced 12 in FIG. 1. Identified is chemical heater section 28 and reaction zone chamber 18. Thermally conductive thick-film metal vias 102 thermally couple the top and bottom sections efficiently for good heat transfer. A thick-film metal layer 104, in intimate contact with chemical heater zone 28 and reaction zone 18, spreads and uniformly heats these sections.

Referring now to FIGs. 5 and 6, illustrated in simplified cross-sectional or plan view is a portion of fuel vaporizer zone 16. More particularly, illustrated in FIG. 5 is a serpentine channel 110, that is typically filled or coated with an inert porous ceramic material for thermal control. Alternatively, vaporizer 16 can be formed of a multiplicity of parallel channels 112, with an inert porous ceramic material for thermal control, in

communication with a porous diffuser 113. In each instance, serpentine channel 110 or parallel channel 112 includes a combined single liquid inlet feed 114 at one end and combining into a single vapor outlet at the other end 116. Serpentine channels 110 and parallel channels 112, are formed in the multi-layer ceramic structure, previously identified as 12. Reaction zone 18 includes similarly designed channels and features as found in the vaporizer zone, discussed above. More particularly, reaction zone 18 includes a reforming catalyst. The catalyst is present as channel wall coatings or as a porous packed bed of catalyst particles. One design goal is to reduce the dimensionality of the reactor channels to the order of a gas diffusion length and control gas residence times, or gas space velocity, that are favorable to reaction kinetics. A multiplicity of parallel reactor channels, generally similar to channels 112, provide for high gas throughput and minimizes back pressure issues as related to channel blockage. It should be understood that it is anticipated by this disclosure that vaporizer zone 16 and reaction zone 18 may include in addition to, or in lieu of, serpentine channels 110 and parallel channels 112, any number of

cavities or chambers, or any combination of channels, cavities, or chambers, thereof.

Illustrated in FIG. 7 in a simplified flow chart diagram, is the chemical reaction 120 that takes place in a steam reforming hydrogen generator, and more particularly in reformer 14 of device 10, as illustrated in FIG. 1. As illustrated, methanol 122 and water 124 are input into a steam reformer 126, generally similar to reformer 14 of FIG. 1. Steam reformer 126 operates at a temperature of approximately 250°C to reform the input methanol 122 and water 124 into a reformed gas mixture, generally referred to as the hydrogen enriched gas. More particularly, in the presence of a catalyst, such as copper oxide, zinc oxide, or copper zinc oxide, the methanol 122 and water 124 solution is reformed into hydrogen, carbon dioxide, and some carbon monoxide. Steam reformer 126 operates in conjunction with an optional carbon monoxide cleanup 128, that in the presence of a preferential oxidation catalyst and air (or  $O_2$ ), reforms a large percentage of the present carbon monoxide into carbon dioxide. This reformed gas mixture supplies fuel through fuel output 130 to an external device, such as a fuel cell.

Accordingly, described is a hydrogen generator including a ceramic carrier defining a reaction zone including a catalyst. The ceramic carrier further includes a heat source thermally coupled to the reaction zone. An inlet channel is supplied for input of the liquid fuel to the fuel processor and an outlet channel is supplied for the output of the reformed gas mixture. The fuel processor device is formed as a monolithically integrated structure, generally comprised of a plurality of thin ceramic layers assembled and then sintered in such a way as to provide for the closed heating zones in which the encapsulated catalysts reforms the inlet fuel into mostly hydrogen gas suitable for supplying fuel via a suitable conduit to an other external device in fluidic communication or integrated monolithically with internal conduits.

While we have shown and described specific embodiments of the present invention, further modifications and improvements will occur to those skilled in the art. We desire it to be understood, therefore, that this invention is not limited to the particular forms shown and we intend

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in the appended claims to cover all modifications that do not depart from the spirit and scope of this invention.



What is claimed is:

1. A hydrogen generator comprising:

a ceramic carrier defining a fuel processor, the fuel processor including a reaction zone including a reforming catalyst;

an inlet channel for liquid fuel; and

an outlet channel for hydrogen enriched gas.

2. A hydrogen generator as claimed in claim 1 wherein the fuel processor further includes a vaporization zone.

3. A hydrogen generator as claimed in claim 1 further including an integrated heat source.

4. A hydrogen generator as claimed in claim 1 wherein the integrated heat source is a resistive heater that is electrically driven.

5. A hydrogen generator as claimed in claim 3 wherein the integrated heater is a chemical heater including a catalyst and arranged to oxidize fuel to produce heat.

6. A hydrogen generator as claimed in claim 5 wherein the chemical heater further includes an air inlet for providing oxygen for the oxidation of the fuel and the inlet channel includes an opening to provide fuel to the chemical heater.

7. A hydrogen generator as claimed in claim 3 wherein the integrated heater couples heat to the reaction zone using thermally conductive structures.

8. A hydrogen generator as claimed in claim 2 wherein one of the vaporization zone and the reaction zone include a plurality of parallel channels.

9. A hydrogen generator as claimed in claim 2 wherein one of the vaporization zone and the reaction zone include a at least one serpentine channel.

10. A hydrogen generator as claimed in claim 1 wherein the ceramic carrier is an monolithic three-dimensional multilayer ceramic structure.

11. A hydrogen generator comprising:

a three-dimensional multilayer ceramic carrier structure defining a fuel reformer including a vaporization zone and a reaction zone including a reforming catalyst;  
an inlet channel for liquid fuel; and  
an outlet channel for hydrogen enriched gas.

12. A hydrogen generator as claimed in claim 11 wherein the ceramic carrier further includes an integrated heater thermally coupled to the reaction zone and the vaporizer zone.

13. A hydrogen generator as claimed in claim 11 wherein the integrated heater is one of a resistive heater that is electrically driven or a chemical heater including a catalyst and arranged to oxidize fuel to produce heat.

14. A hydrogen generator as claimed in claim 13 wherein the chemical heater further includes an air port for providing oxygen for the oxidation of the fuel and the

inlet channel includes an opening to provide fuel to the chemical heater.

15. A hydrogen generator as claimed in claim 11 wherein the integrated heater couples heat to the reaction zone using thermally conductive structures.

16. A hydrogen generator as claimed in claim 11 wherein one of the vaporization zone and the reaction zone include a plurality of parallel structures.

17. A hydrogen generator as claimed in claim 11 wherein one of the vaporization zone and the reaction zone include at least one serpentine channel.

18. A hydrogen generator comprising:

a three-dimensional multilayer ceramic carrier structure defining a fuel processor including a vaporization zone and a reaction zone including a reforming catalyst, at least one of the vaporization zone and the reaction zone including one of a plurality of parallel channels or at least one serpentine channel, the ceramic

carrier further including an integrated heater thermally coupled to the reaction zone and the vaporization zone using thermally conductive structures;

an inlet channel for liquid fuel; and

an outlet channel for hydrogen enriched gas.

19. A hydrogen generator as claimed in claim 18 further including an integrated heater thermally coupled to the reaction zone using thermally conductive channels.

20. A hydrogen generator as claimed in claim 18 wherein the integrated heater is one of a resistive heater that is electrically driven or a chemical heater including a catalyst and arranged to oxidize fuel to produce heat.

21. A hydrogen generator as claimed in claim 20 wherein the chemical heater further includes an air port for providing oxygen for the oxidation of the fuel and the inlet channel includes an opening to provide fuel to the chemical heater.

HYDROGEN GENERATOR UTILIZING CERAMIC TECHNOLOGY

Abstract of the Disclosure

A hydrogen generator including a three-dimensional multilayer ceramic carrier structure defining a fuel reformer. The reformer includes a vaporization zone and a reaction zone including a catalyst. The reformer is operational as either a steam reformer, a partial oxidation reformer or an autothermal reformer. The fuel reformer, or processor, further includes an inlet channel for liquid fuel and an outlet channel for hydrogen enriched gas. The fuel processor is formed utilizing multi-layer ceramic technology in which thin ceramic layers are assembled then sintered to provide miniature dimensions in which the encapsulated catalyst converts or reforms inlet fuel into a hydrogen enriched gas.



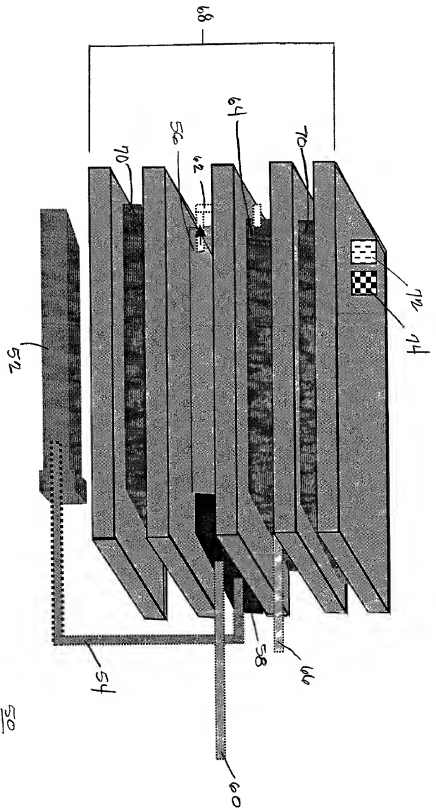
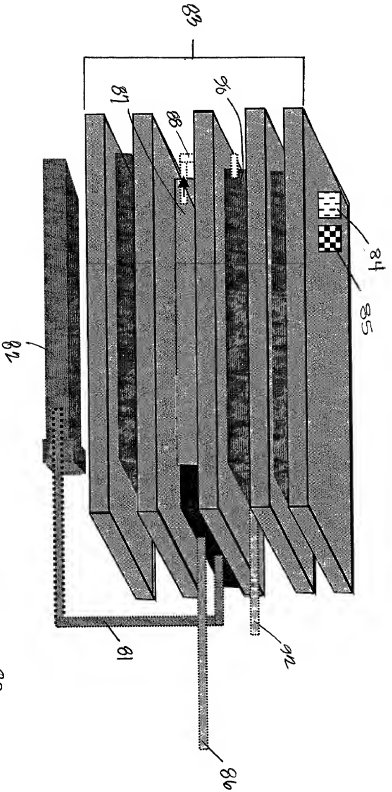


FIG. 2





80  
FIG. 3

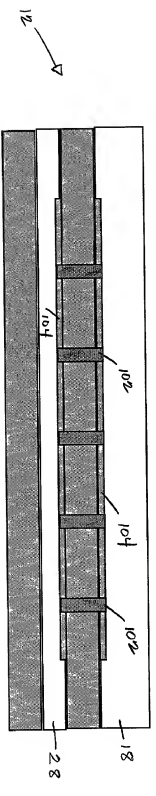


FIG. 4

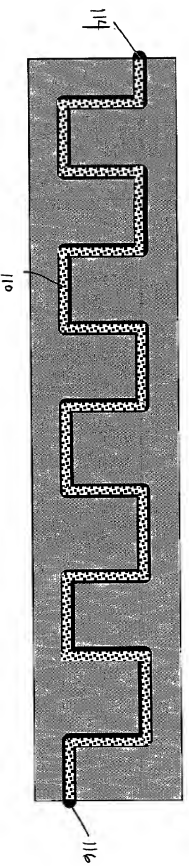


FIG. 5

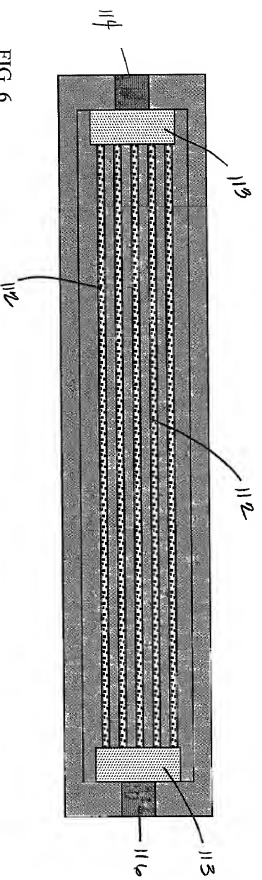


FIG. 6

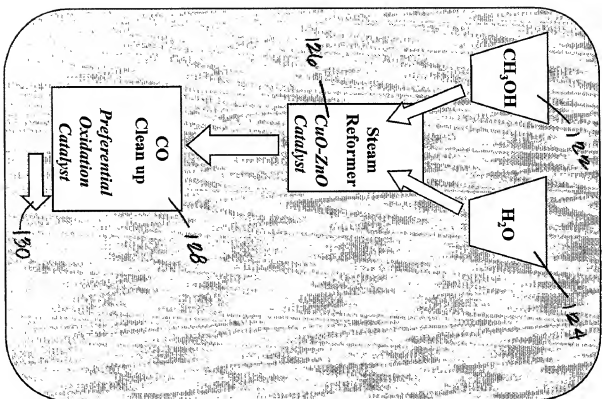


FIG. 7

**COMBINED DECLARATION AND POWER OF ATTORNEY  
FOR PATENT APPLICATION**

Attorney Docket CT00-013

As a below named inventor, I hereby declare that:

My residence, post office address and citizenship are as stated below next to my name.

I believe I am the original, first and sole inventor (if only one name is listed below), or an original, first and joint inventor (if plural names are listed below), of the subject matter which is claimed and for which a patent is sought on the invention entitled HYDROGEN GENERATOR UTILIZING CERAMIC TECHNOLOGY, the specification of which is attached hereto unless the following box is checked:

☐ Application was filed on \_\_\_\_\_  
as Application No. \_\_\_\_\_  
and was amended on \_\_\_\_\_.

I hereby state that I have reviewed and understand the contents of the above identified specification, including the claims, as amended by any amendment referred to above.

I acknowledge the duty to disclose information which is material to the patentability of this application in accordance with Title 37, Code of Federal Regulations, §1.56.

I hereby claim foreign priority benefits under Title 35, United States Code, § 119(a)-(d) or 365(b) any foreign application(s) for patent or inventor's certificate, or 365(a) of any PCT international application which designated at least one country other than the United States of America, listed below and have also identified below, any foreign application for patent or inventor's certificate, or of any PCT international application having a filing date before that of the application on which priority is claimed.

Prior Foreign Application(s)		Priority Claimed
_____	_____	<input type="checkbox"/> Yes <input type="checkbox"/> No
(Number)	(Country)	(Day/Month/Year Filed)
_____	_____	<input type="checkbox"/> Yes <input type="checkbox"/> No
(Number)	(Country)	(Day/Month/Year Filed)

I hereby claim the benefit under Title 35, United States Code, § 119 of any United States provisional application(s), listed below:

_____	_____
(Application Number)	(Filing Date)
_____	_____
(Application Number)	(Filing Date)

I hereby claim the benefit under Title 35, United States Code, § 120 of any United States application(s), or 365(c) of any PCT international application designating the United States of America, listed below:

(U.S. Parent Application Number or PCT Parent No.)	(Filing Date)	(Country)
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I hereby appoint the attorney(s) and/or agent(s) associated with Customer Number 23330 to prosecute this application and to transact all business in the Patent and Trademark Office connected therewith.

Address all telephone calls to Mr. William E. Koch at telephone no. (480) 441-4281.

Address all correspondence to customer number 23330.

I hereby declare that all statements made herein of my own knowledge are true and that all statements made on information and belief are believed to be true; and further that these statements were made with the knowledge that willful false statements and the like so made are punishable by fine or imprisonment, or both, under Section 1001 of Title 18 of the United States Code and that such willful false statements may jeopardize the validity of the application or any patent issued thereon.

FULL NAME OF FIRST INVENTOR: FIRST MIDDLE LAST		INVENTOR'S SIGNATURE	DATE (SPELL OUT MONTH)
Chowdary Ramesh Koripella		<i>Chowdary R Koripella</i>	Aug 23, 2000
RESIDENCE: 11308 N. 118th Street, Scottsdale, AZ 85259		CITIZENSHIP: United States	
POST OFFICE ADDRESS: Same as above			

FULL NAME OF SECOND INVENTOR: FIRST MIDDLE LAST		INVENTOR'S SIGNATURE	DATE (SPELL OUT MONTH)
Christopher K. Dyer		<i>Christopher K Dyer</i>	Aug 23, 2000
RESIDENCE: 306 Roger Avenue, Westfield, NJ 07090		CITIZENSHIP: United Kingdom	
POST OFFICE ADDRESS: Same as above			

FULL NAME OF THIRD INVENTOR: FIRST MIDDLE LAST			INVENTOR'S SIGNATURE:	DATE: (SPELL OUT MONTH)
Stephen P. Rogers			<i>Stephen P. Rogers</i>	8/23/00
RESIDENCE:			CITIZENSHIP:	
516 E. Mountain Sky Avenue, Phoenix, AZ 85048			United States	
POST OFFICE ADDRESS:				
Same as above				

FULL NAME OF FOURTH INVENTOR: FIRST MIDDLE LAST			INVENTOR'S SIGNATURE:	DATE: (SPELL OUT MONTH)
Dominic Francis Gervasio			<i>Dominic Francis Gervasio</i>	8-23-2000
RESIDENCE:			CITIZENSHIP:	
14406 North Century Drive, Fountain Hills, AZ 85268			United States	
POST OFFICE ADDRESS:				
Same as above				